

CLAIMS

1. A method of decontaminating a gas intended for use in photolithography
2 and metrology to remove from the gas or reduce the concentration in the gas of
a contaminant which interferes with light transmittance, which comprises
4 removing said contaminant from the gas by passing the gas through a body of
decontaminant comprising 10 to 80 percent by volume of an electropositive
6 metal component; 10 to 80 percent by volume of a high silica zeolite; and 10 to
80 percent by volume of a late transition metal compound, wherein each
8 component comprises at least 10 percent by volume of said composition and the
total of the components equals 100 percent by volume.
2. A method as in Claim 1 wherein the body of the decontaminant
2 comprises a mixture of an electropositive metal component, a high silica zeolite,
and a late transition metal compound in a ratio on or within the area bounded by
4 the line A-B-C in Figure 2.
3. A method as in Claim 1 further comprising removal of the contaminant
2 prior to or after treatment of the gas to remove solid particulate matter from the
gas.
4. A method as in Claim 1 wherein the contaminant is gaseous.
5. A method as in Claim 4 wherein the contaminant comprises a neutral
2 polar gaseous molecule.
6. A method as in Claim 4 wherein the contaminant comprises a neutral
2 polar aprotic molecule.

- 2 7. A method as in Claim 4 wherein the contaminant comprises a protic or
aprotic alkaline molecule.
- 2 8. A method as in Claim 4 wherein the contaminant comprises an acidic
2 polar species.
- 2 9. A method as in Claim 4 wherein the contaminant comprises an
2 environmental gas.
- 2 10. A method as in Claim 1 wherein the contaminant comprises at least one
2 of gaseous or entrained water, an alcohol, a nitrogen oxide, a sulfur oxide, an
organic sulfide, an organic halide, an amine, a hydrocarbon, a siloxane, a
4 carbon oxide or an environmental gas.
- 2 11. A method as in Claim 1 wherein the gas after decontamination is used in
2 photolithography.
- 2 12. A method as in Claim 1 wherein the gas after decontamination is used in
2 metrology.
13. A method as in Claim 1 wherein the concentration of the contaminant in
the gas is reduced to not more than 1 ppb.
- 2 14. A method as in Claim 13 wherein the concentration of the contaminant
2 in the stream is reduced to not more than 100 ppt.
- 2 15. A method as in Claim 14 wherein the concentration of the contaminant
2 in the gas is reduced to not more than 1 ppt.

16. A method as in Claim 1 wherein the contaminant removed or reduced comprises a contaminant which interferes with transmittance of light having a wavelength of 248 nm.

2 17. A method as in Claim 16 wherein the concentration of the contaminant in the gas is reduced to not more than 1 ppb.

2 18. A method as in Claim 1 wherein the contaminant removed or reduced comprises a contaminant which interferes with transmittance of light having a wavelength of 193 nm.

2 19. A method as in Claim 18 wherein the concentration of the contaminant in the gas is reduced to not more than 100 ppt.

2 20. A method as in Claim 1 wherein the contaminant removed or reduced comprises a contaminant which interferes with transmittance of light having a wavelength of 157 nm.

2 21. A method as in Claim 20 wherein the concentration of the contaminant in the gas is reduced to not more than 100 ppt.

2 22. A method as in Claim 1 wherein the contaminant removed or reduced comprises a contaminant which interferes with transmittance of light having a wavelength ≤ 100 nm.

2 23. A method as in Claim 22 wherein the concentration of the contaminant in the gas is reduced to not more than 1 ppt.

2 24. A method as in Claim 1 further comprising a providing a generator of coherent light as a source of light for the photolithography or metrology.

25. A method as in Claim 24 wherein the generator or coherent light
2 comprises a laser.

26. A method as in Claim 1 further comprising a providing a generator of
2 noncoherent light as a source of light for the photolithography or metrology.

27. A composition for decontaminating a gas intended for use in
2 photolithography and metrology by removing from the gas or reducing the
concentration in the gas of a contaminant which interferes with light
4 transmittance, which composition comprises:

10 to 80 percent by volume of an electropositive metal component;
6 10 to 80 percent by volume of a high silica zeolite; and
10 to 80 percent by volume of a late transition metal compound,
8 wherein the total composition comprises 100 percent by volume.

28. A composition as in Claim 27 which effects a reduction in the
2 contaminant content of said gas stream to not more than 1 ppb.

29. A composition as in Claim 28 which effects a reduction in the contaminant
2 content of said gas stream to not more than 100 ppt.

30. A composition as in Claim 29 which effects a reduction in the contaminant
2 content of said gas stream to not more than 1 ppt.

31. A composition as in Claim 27 wherein at least one of the electropositive
2 metal component, the high silica zeolite and the late transition metal compound
is in a particulate, coating, pelleted, extruded, plate or powder form, or a mixture
4 of such forms.

2 32. A composition as in Claim 27 wherein the electropositive metal in the electropositive metal component comprises a Group 3 or 4 metal, metal salt or metal oxide.

2 33. A composition as in Claim 32 wherein said electropositive metal component comprises a Group 3 or 4 metal oxide selected from the group consisting of titania, zirconia, yttria, or vanadia.

34. A composition as in Claim 33, wherein the metal oxide comprises titania.

2 35. A composition as in Claim 32, wherein the electropositive metal component has a surface area in the range of 140-1200 m²/g.

2 36. A composition as in Claim 35 wherein the electropositive metal component has a surface area in the range of 140-500 m²/g.

2 37. A composition as in Claim 27, wherein said high silica zeolite comprises a zeolitic structure with an silica:alumina ratio of at least 90:1.

2 38. A composition as in Claim 37 wherein said high silica zeolite comprises a zeolitic structure and a silica:alumina ratio of at least 300:1.

2 39. A composition as in Claim 38 wherein said high silica zeolite comprises a zeolitic structure and a silica:alumina ratio of at least 400:1.

2 40. A composition as in Claim 39 wherein said high silica zeolite comprises a zeolitic structure and a silica:alumina ratio in the range of 400-2000:1.

2 41. A composition as in Claim 27 wherein said high silica zeolite comprises Zeolite Y or zeolite ZSM-5.